

UNIVERSITY OF JAFFNA, SRI LANKA FACULTY OF ALLIED HEALTH SCIENCES FIRST YEAR SECOND SEMESTER EXAMINATION IN BScHons (MLS) - 2023 MLSMT 1213 MEDICAL LABORATORY TECHNOLOGY II

Date: 20.06.2025 Time: 2 hours

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Answe	r all <u>Six</u> Questions.	
1. 1.1	Describe the working principle of UV/Visible spectroscopy.	(20 Marks)
1.2	Explain the causes and corrective actions of the following interference	es that may
	occur in atomic absorption spectroscopy.	
	1.2.1 Spectral interferences	(20 Marks)
	1.2.2 Chemical interferences	(20 Marks)
	1.2.3 Ionization interference	(20 Marks)
1.3 5	State the clinical applications of Nephelometry.	(20 Marks)
2. 2.1	Outline the principle of Hydrophobic Interaction chromatography.	(20 Marks)
2.2	Explain the steps of performing a Hydrophobic Interaction chromato	graphy to purify
	a newly invented drug molecule.	(60 Marks)
2.3	Give the applications of gel permeation chromatography.	(20 Marks)
3. 3.1	Briefly describe the scientific basis of DNA gel electrophoresis.	(20 Marks)
3.2	Explain the usage of the following in a Protein gel electrophoresis.	
	3.2.1. Bromophenol blue	(20 Marks)
	3.2.2. Coomassie brilliant blue	(20 Marks)
	3.2.3. Glycerol	(20 Marks)
3.3	State the importance of using Ethidium Bromide in DNA Gel Electro	phoresis.
		(20 Marks)

4.	4.1	Describe the principle of mass spectrometry.	(50 marks)
	4.2	Name four (4) different types of mass analyzers used in mass spectromet	ry.
			(20 marks)
	4.3	Outline the steps of measuring serum electrolytes (Na+ and K+) in flame	photometry.
			(30 Marks)
5.	5.1	Describe the working principle of,	
		5.1.1. Anion exchange chromatography.	(20 Marks)
		5.1.2. Paper chromatography	(20 Marks)
	5.2	Explain how thin-layer chromatography (TLC) can be used to identify the	e presence
	of a	an amino acid impurity in a purified drug sample	(60 Marks)
6.	W	Trite short notes on	
	6.1	pH electrode.	(25 Marks)
	6.2	Clark-style amperometric O ₂ sensor.	(25 Marks)
	6.3	Pneumatic tube systems.	(25 Marks)
	6.4	Ion-Selective Electrodes.	(25 Marks)